

INFRARED STUDY ON THE C≡P STRETCHING VIBRATION OF 1-PHOSPHAPROPYNE
CH₃C≡P AND ITS PERDEUTERIUM ANALOG CD₃C≡P

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Unstable molecule 1-phosphapropyne was detected by gas-phase infrared spectroscopy and the vibration-rotation band of the C≡P stretching was analyzed for CH₃C≡P and CD₃C≡P.

In 1979, an unstable molecule 1-phosphapropyne CH₃C≡P was first found by microwave and photoelectron spectroscopy.¹⁾ This initiative work stimulated us to undertake vibration-spectroscopic investigation of this molecule. Mixture samples containing CH₃C≡P were prepared by the pyrolysis of C₂H₅PCl₂¹⁾ and the infrared bands of CH₃C≡P were assigned on the basis of the spectral changes with time and of the observed band shapes.²⁾ In this work, we have dealt with the vibration-rotation analysis of the C≡P stretching band for CH₃C≡P and CD₃C≡P in order to make definite band assignment of 1-phosphapropyne.

The infrared spectra were measured on a JEOL JIR-40X Fourier transform infrared spectrometer with a resolution of 0.08 cm⁻¹. The sample pressure was about 3 kPa in a 12-cm glass cell. In either of the infrared spectra of the pyrolysis products of C₂H₅PCl₂ and C₂D₅PCl₂, a strong parallel band was observed at 1550-1560 cm⁻¹ as shown in Fig. 1. This band exhibited a strong Q branch as well as series of P and R branches with line spacing of about 2B; possible K structures were not resolved under the spectral resolution available. The J-value assignment of the P- and R-branch lines were readily made, as indicated in Fig. 1, up to J=60 for CH₃C≡P and J=80 for CD₃C≡P. In the spectral analysis, the ground-state constants were first determined by the combination differences method.³⁾ The agreement between the derived value for B₀ and the value obtained by microwave spectroscopy¹⁾ indicated that the parallel band at 1550-1560 cm⁻¹ was indeed due to 1-phosphapropyne. This band was thus assigned to the C≡P stretching mode (ν₂). The spectroscopic constants for the ν₂ band were subsequently derived from the observed vibration-rotation wavenumbers by constraining the ground-state constants B₀, (D_J)₀, and (D_{JK})₀ to the accurately determined microwave values¹⁾ and assuming (D_{JK})₂=(D_{JK})₀, (D_K)₂=(D_K)₀, and [(A₂-B₂)-(A₀-B₀)]=0. In assigning the quantum number K to each of the unresolved P(J) and R(J) lines, spectral simulation with the Boltzmann factor was employed. Table 1 gives the spectroscopic constants determined for the ν₂ band of CH₃C≡P and CD₃C≡P. The standard errors of the least-squares fit of 124 lines for CH₃C≡P and of 178 lines for CD₃C≡P were ±0.020 and ±0.023 cm⁻¹, respectively.

The fact that the ν₂ wavenumbers for CH₃C≡P and CD₃C≡P are almost the same indicates that the C≡P stretching vibration couples negligibly with methyl group

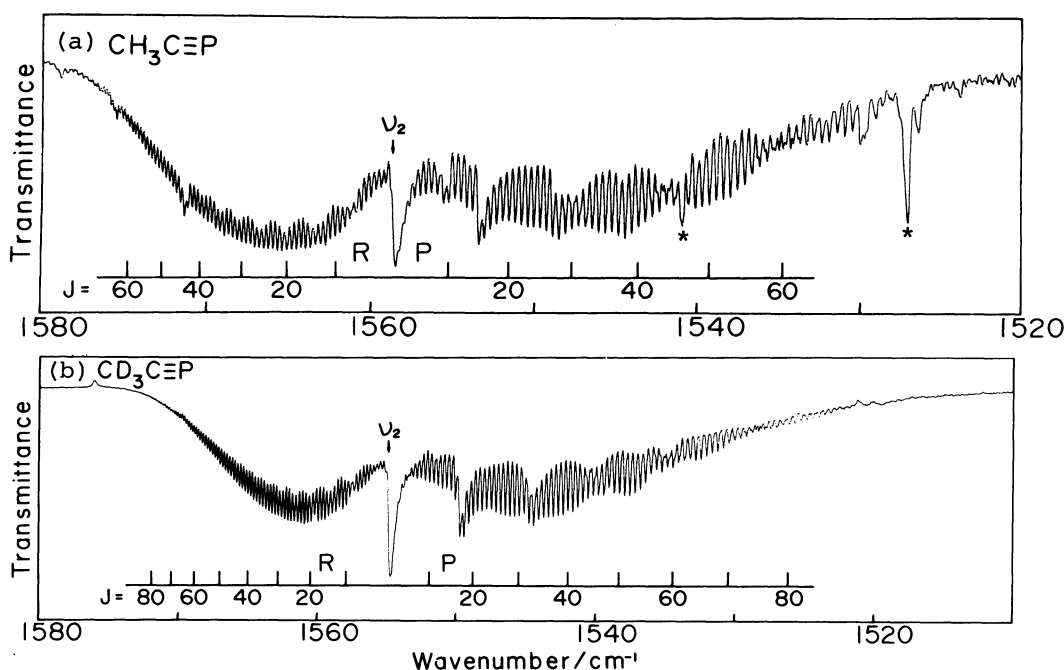


Fig. 1. Vibration-rotation spectra of the ν_2 band region of $\text{CH}_3\text{C}\equiv\text{P}$ (a) and $\text{CD}_3\text{C}\equiv\text{P}$ (b). The peaks marked with asterisks are R_{Q_K} branches of the ν_6 band.

vibrations. This contrasts with the large coupling of the $\text{C}\equiv\text{P}$ stretching vibration with the skeletal stretching vibration; the observed $\text{C}\equiv\text{P}$ stretching wavenumbers are in fact spread over in the $1270\text{--}1680\text{ cm}^{-1}$ region for $\text{HC}\equiv\text{P}$, $\text{FC}\equiv\text{P}$, and $\text{CH}_3\text{C}\equiv\text{P}$.²⁾

The Q branches of the hot bands, $\nu_2+\nu_8-\nu_8$ and $\nu_2+2\nu_8-2\nu_8$, were also observed at 1553.63 and 1548.67 cm^{-1} , respectively, for $\text{CH}_3\text{C}\equiv\text{P}$ and at 1549.91 and 1544.91 cm^{-1} for $\text{CD}_3\text{C}\equiv\text{P}$. The anharmonicity constant x_{28} was therefore derived to be $-5.03(7)\text{ cm}^{-1}$ for either of the isotopic species.

In conclusion, the parallel band at $1550\text{--}1560\text{ cm}^{-1}$ was assigned to the $\text{C}\equiv\text{P}$ stretching vibration of 1-phosphapropyne, on the basis of the derived value for the rotational constant B and the small wavenumber shift of this band on deuteration.

Table 1. Spectroscopic constants (cm^{-1}) of the ν_2 band for $\text{CH}_3\text{C}\equiv\text{P}$ and $\text{CD}_3\text{C}\equiv\text{P}$ ^{a)}

Constant	$\text{CH}_3\text{C}\equiv\text{P}$		$\text{CD}_3\text{C}\equiv\text{P}$	
	ν_2	Ground state	ν_2	Ground state
$(\nu_0)_v$	1558.724(25)	-	1554.962(25)	-
B_v	0.165368(15)	0.16649312 ^{b)}	0.1423070(74)	0.14322105 ^{b)}
$(D_J)_v \times 10^8$	2.45(43)	2.74 ^{b)}	2.53(10)	2.67 ^{b)}
$(D_{JK})_v \times 10^6$	$= (D_{JK})_0$	2.2212 ^{b)}	$= (D_{JK})_0$	1.4343 ^{b)}

a) Errors (3σ) for the last significant figures in parentheses. b) Microwave value.

References

- 1) H. W. Kroto, J. F. Nixon, and N. P. C. Simmons, *J. Mol. Spectrosc.*, **77**, 270 (1979); N. P. C. Westwood, H. W. Kroto, J. F. Nixon, and N. P. C. Simmons, *J. Chem. Soc., Dalton Trans.*, **1979**, 1405.
- 2) K. Ohno, H. Matsuura, and H. Murata, *J. Phys. Chem.*, in press.
- 3) H. C. Allen, Jr. and P. C. Cross, "Molecular Vib-Rotors," John Wiley and Sons, Inc. New York (1963).

(Received January 11, 1984)